

Time dependence of expectation value $\frac{d\langle \hat{A} \rangle}{dt} = ?$

$$\hat{H} |\psi\rangle = E |\psi\rangle$$

If time dependent:

$$\hat{H} |\psi\rangle = i\hbar \frac{\partial}{\partial t} |\psi\rangle \rightarrow \left| \frac{\partial \psi}{\partial t} \right\rangle = \frac{-i}{\hbar} \hat{H} |\psi\rangle$$

$$\left\langle \frac{\partial \psi}{\partial t} \right| = \frac{i}{\hbar} \langle \psi | \hat{H}$$

$$\frac{d\langle \hat{A} \rangle}{dt} = \frac{d}{dt} \langle \psi | \hat{A} | \psi \rangle$$

$$= \left\langle \frac{\partial \psi}{\partial t} \right| \hat{A} | \psi \rangle + \langle \psi | \frac{\partial \hat{A}}{\partial t} | \psi \rangle + \langle \psi | \hat{A} \left| \frac{\partial \psi}{\partial t} \right\rangle$$

$$= \frac{i}{\hbar} \langle \psi | \hat{H} \hat{A} | \psi \rangle - \frac{i}{\hbar} \langle \psi | \hat{A} \hat{H} | \psi \rangle + \langle \psi | \frac{\partial \hat{A}}{\partial t} | \psi \rangle$$

$$= \frac{i}{\hbar} \langle \psi | \hat{H} \hat{A} - \hat{A} \hat{H} | \psi \rangle + \langle \psi | \frac{\partial \hat{A}}{\partial t} | \psi \rangle$$

$$= \frac{i}{\hbar} \langle \psi | [\hat{H}, \hat{A}] | \psi \rangle + \left\langle \frac{\partial \hat{A}}{\partial t} \right\rangle$$

$$\frac{d}{dt} \langle \hat{A} \rangle = \frac{i}{\hbar} \langle [\hat{H}, \hat{A}] \rangle + \left\langle \frac{\partial \hat{A}}{\partial t} \right\rangle$$

So if \hat{A} is time independent:

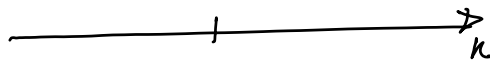
$$\frac{d}{dt} \langle \hat{A} \rangle = \frac{i}{\hbar} \langle [\hat{H}, \hat{A}] \rangle$$

Easier to remember if write it in this form:

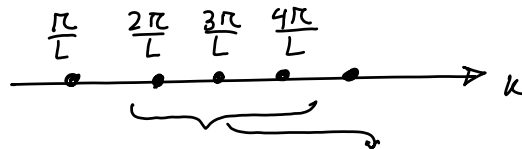
$$[H, A] = -i\hbar \frac{dA}{dt}$$

DENSITY OF STATES

Free electron $E = \frac{\hbar^2 k^2}{2m}$ all k 's are acceptable

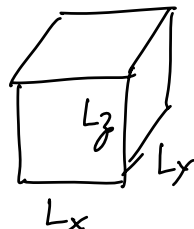


Electron in 1D $E = \frac{\hbar^2 k_x^2}{2m}$ k_x are quantized:
 $k_x = \frac{n\pi}{L}$ $n = 1, 2, \dots$



So for example, within this size of k ,
 we only have three states for the
 electron.

Particle in a box

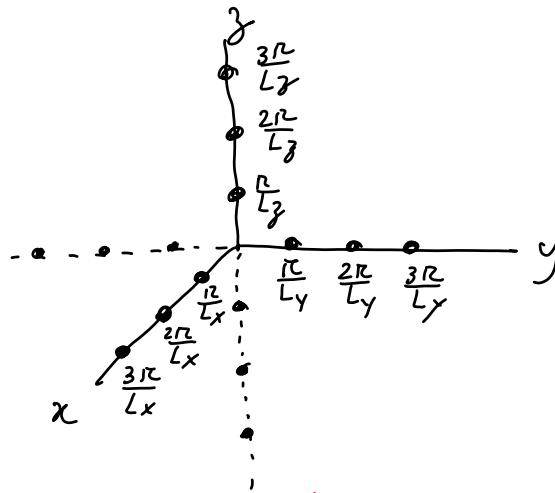


$$E = \frac{\hbar^2 k_x^2}{2m} + \frac{\hbar^2 k_y^2}{2m} + \frac{\hbar^2 k_z^2}{2m}$$

$$k_x = \frac{n_x \pi}{L_x} \quad n_x = 1, 2, \dots$$

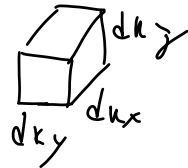
$$k_y = \frac{n_y \pi}{L_y} \quad n_y = 1, 2, 3, \dots$$

$$k_z = \frac{n_z \pi}{L_z} \quad n_z = 1, 2, 3, \dots$$



Question: How many states are available
in a unit volume? or: what is
the density of states in k -space?

The volume is: $dk_x dk_y dk_z$



The volume for each state is: $\left(\frac{2\pi}{L_x}\right) \left(\frac{2\pi}{L_y}\right) \left(\frac{2\pi}{L_z}\right) \boxplus$

So the number of states in volume dk is:

$$\frac{dk_x dk_y dk_z}{\frac{(2\pi)^3}{L_x L_y L_z}} = V \frac{dk_x dk_y dk_z}{(2\pi)^3}$$

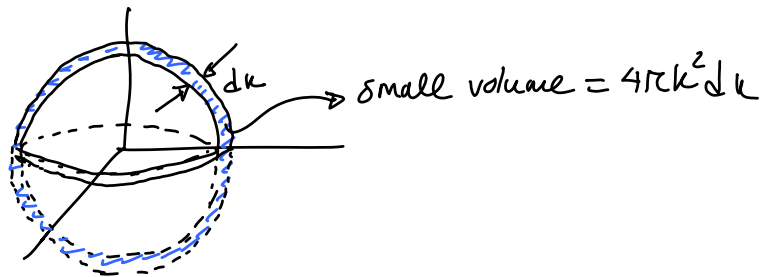
V : volume \rightarrow

So the # of states per unit volume is:

$$N(k) dk_x dk_y dk_z = \frac{dk_x dk_y dk_z}{(2\pi)^3} \times 2$$

↑
Spin up & down

In spherical coordinate: $dk_x dk_y dk_z = 4\pi k^2 dk$



$$N(k) dk = \frac{1}{(2\pi)^3} 4\pi k^2 dk \times 2$$

↑
spin

We usually like to know the density of states

in Energy. $N(E) = ?$

$$N(E) dE = N(k) \frac{dk}{dE} dE \quad \text{so we should calculate } \frac{dk}{dE}$$

$$\text{But } E = \frac{\hbar^2 k^2}{2m} \Rightarrow dE = \frac{\hbar^2}{m} k dk \Rightarrow \frac{dk}{dE} = \frac{m}{\hbar^2 k} \Rightarrow$$

$$N(E) dE = 2 \times \frac{4\pi k^2}{(2\pi)^3} \frac{m}{\hbar^2 k} dE$$

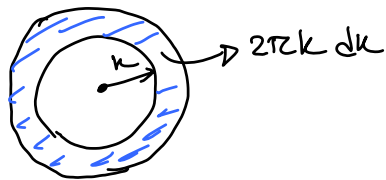
spin
↓

$$N(E) = \frac{2m}{2\pi^2 \hbar^2} \frac{1}{k} = \frac{m}{\pi^2 \hbar^2} \frac{\sqrt{2mE}}{\hbar}$$

$$N(E) = \frac{2^{1/2} m^{3/2}}{\pi^2 \hbar^3} \sqrt{E}$$

Energy DOS in 3D

In 2D:



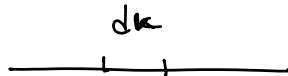
$$N(E) dE = \overset{\text{spin}}{\downarrow} 2 \frac{2\pi k dk}{(2\pi)^2}$$

$$N(E) = \frac{1}{\pi} k \frac{dk}{dE} = \frac{1}{\pi} k \frac{m}{\hbar^2 k} = \frac{m}{\pi \hbar^2}$$

$$\overset{2D}{N(E)} = \frac{m}{\pi \hbar^2}$$

In 1D

spin



$$N(E) dE = \overset{\text{spin}}{\downarrow} 2 \frac{dk}{2\pi}$$

$$N(E) = \frac{1}{\pi} \frac{dk}{dE} = \frac{1}{\pi} \frac{m}{\hbar^2 k} = \frac{m}{\pi \hbar^2} \frac{\hbar}{\sqrt{2mE}}$$

$$\overset{1D}{N(E)} = \frac{m^{1/2}}{2^{1/2} \pi \hbar} \frac{1}{\sqrt{E}}$$

